

New cementitious binders for immobilization of radioactive waste

EURECA-PRO

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KNOWLEDGE IN ACTION

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NuTeC

Nuclear Technologisch Centrum

Centre for Environmental Science (CMK)



CMK conducts fundamental and applied research:

1. To understand **influences of the environment** on organisms



2. To develop and assess sustainable **clean technologies** to mitigate influences of the environment on organisms



3. To monitor, value and **optimize biodiversity and ecosystem services** under different stress conditions, including climate change



NuTeC – Research in environmental technology

▪ Nuclear decommissioning and immobilization of nuclear waste



Characterization

Treatment

Encapsulation -immobilization

▪ Recycling of naturally occurring radioactive materials (NORM)



Characterization

Treatment

Recycling in construction

▪ Adsorption & desorption of (radiological) pollutants



Pyrolysis technology

Adsorbent Biochar, AC, zeolite, ...

Adsorption/desorption of pollutants in water, air, soil

Outline

1. Introduction – current situation

2. New cementitious binders

- a) Immobilization
- b) Effects of ionizing radiation on the encapsulation matrix
- c) Treatment/management of liquid radioactive waste

3. Lessons learned

Classification radioactive waste (Belgium)

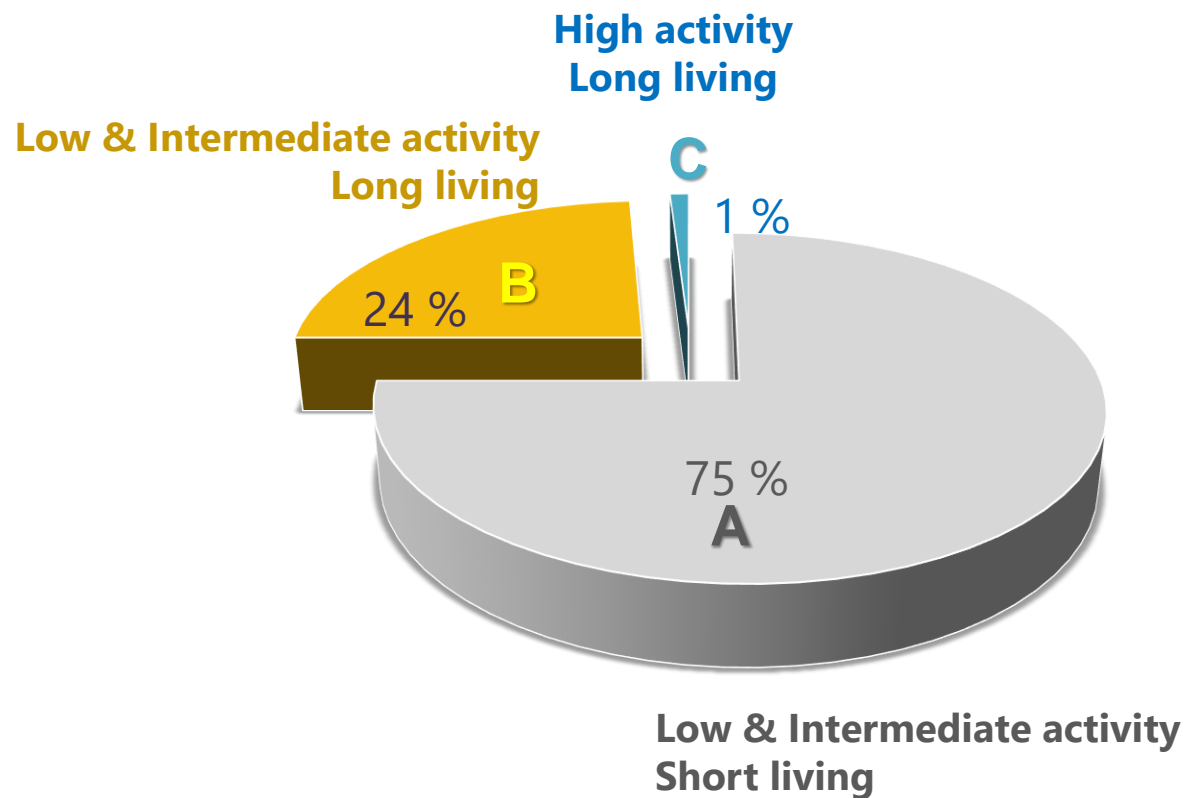
- Category **A** – **B** – **C** waste

	Low activity [< 5 mSv/h]	Intermediate activity	High Activity [> 2 Sv/h]
Short half life [< 30 year]	A	A	C
Long half life [> 30 year]	B	B	C



Classification radioactive waste (Belgium)

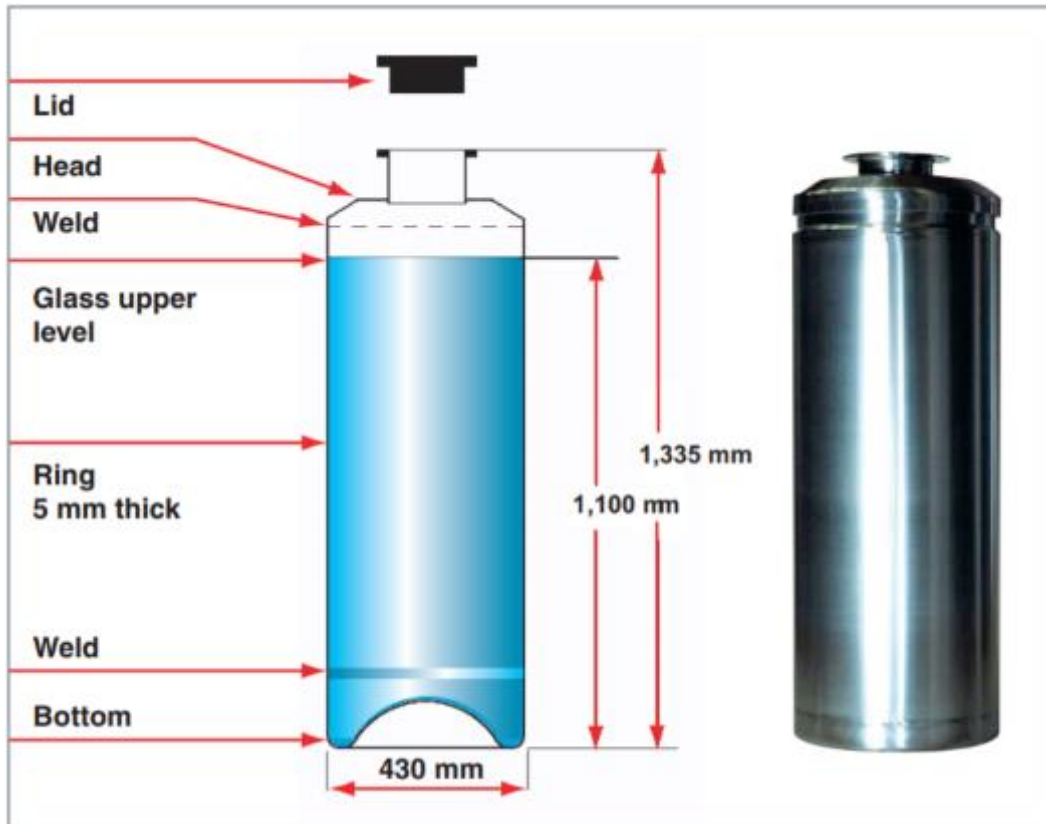
- Category A – B – C waste



High activity waste: vitrification

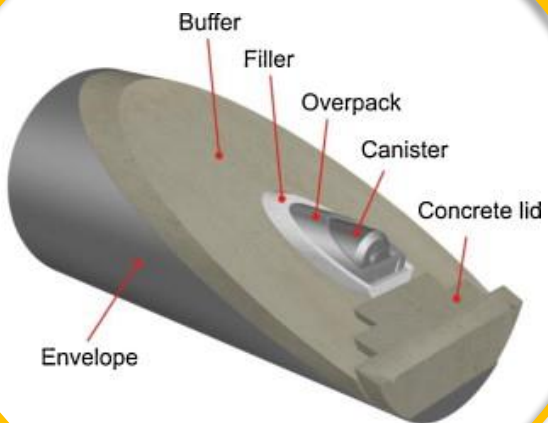


- Liquid waste added to **ceramic- or glass-**forming chemicals, mixed and thermally treated to immobilize in solid waste-forms.
- **Effective** but very **expensive** → less suitable for larger volumes

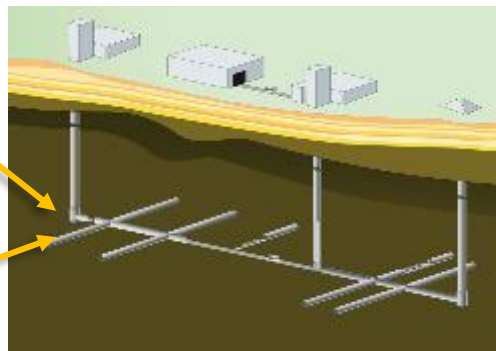


High-level waste: Deep geological disposal

C



B & C



Low and Intermediate Level Waste: cementation

- Widely used: compaction or incineration, followed by **cementation** as a cost-efficient and proven technology



Short living waste (< 30 year)

Low and Intermediate Level Waste: near surface disposal



Low and Intermediate Level Waste: cementation

Current
methods

Cementation using ordinary Portland cement (OPC)

Advantages
OPC

Availability

Relative **low-cost**

Easily modified to fit specific purpose

Robust and tolerant to a wide variety of wastes

Relatively easily **processed remotely**

...

Low and Intermediate Level Waste: cementation

Current
methods

Cementation using ordinary Portland cement (OPC)

Disadvantages
OPC ^a

***High temperature increase** resulting from the heat of hydration, giving rise to thermal stresses and possible cracking.

*The **weak immobilization potential** for monovalent cations such as Cs⁺.

***Moderate durability** (e.g. leaching through alkali-silica reaction (ASR) with certain types of RAW under certain conditions)

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Similar materials, different uses

Construction

- High strength
- Fast strength development
- Control of flow properties with organic admixtures
- Durability for 50-200 years service life
- Passivation of mild steel
- Low cost

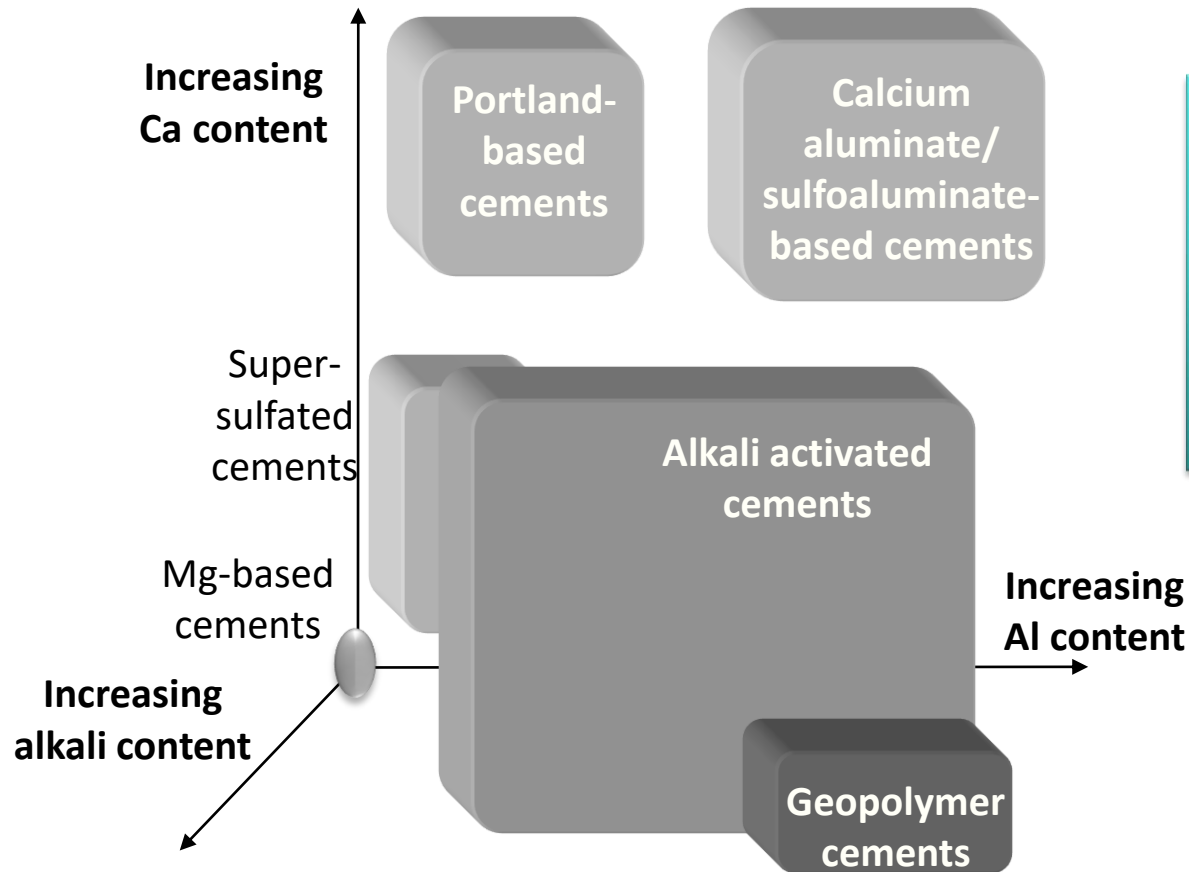
Waste immobilisation

- **Durability** for 100,000 – 1,000,000 years service life
- **Binding of radionuclides**
- **Low heat evolution**
- **Dimensional stability**
- Controlled **corrosion** of reactive metals
- Stability under **irradiation**

Can we realistically expect all this from a single material?



Traditional and non-traditional cements



Moving from a single universal cement to an array of cement types

Most important: Designing materials that are fit for purpose!

Production Alkali activated materials (AAMs)

Solid **aluminosilicate** source + **Alkali** silicate/hydroxide **activating** solution

Precursor

Dissolution
Oligomerization
Polymerization

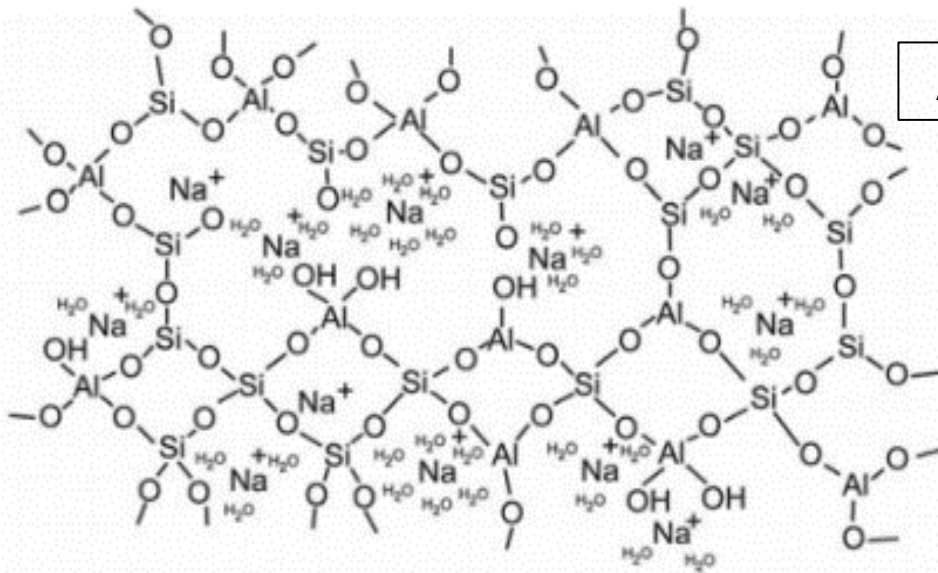
Activator

Synthesis parameters

Aluminosilicate polymer

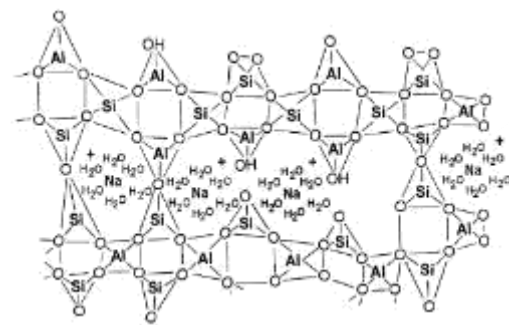
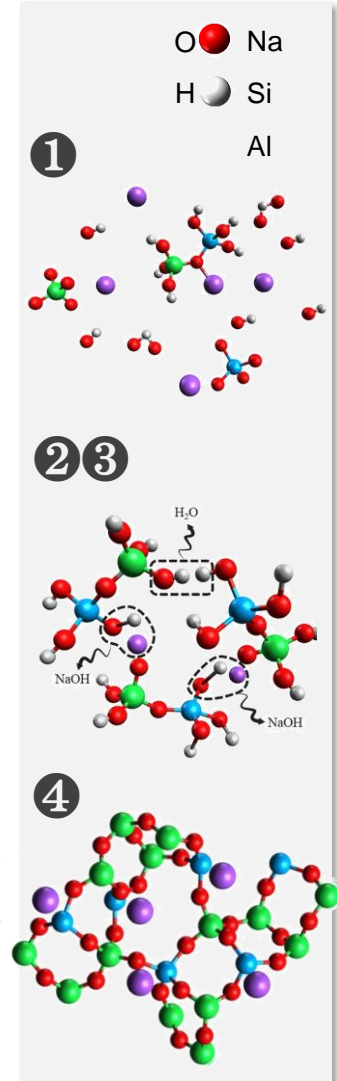
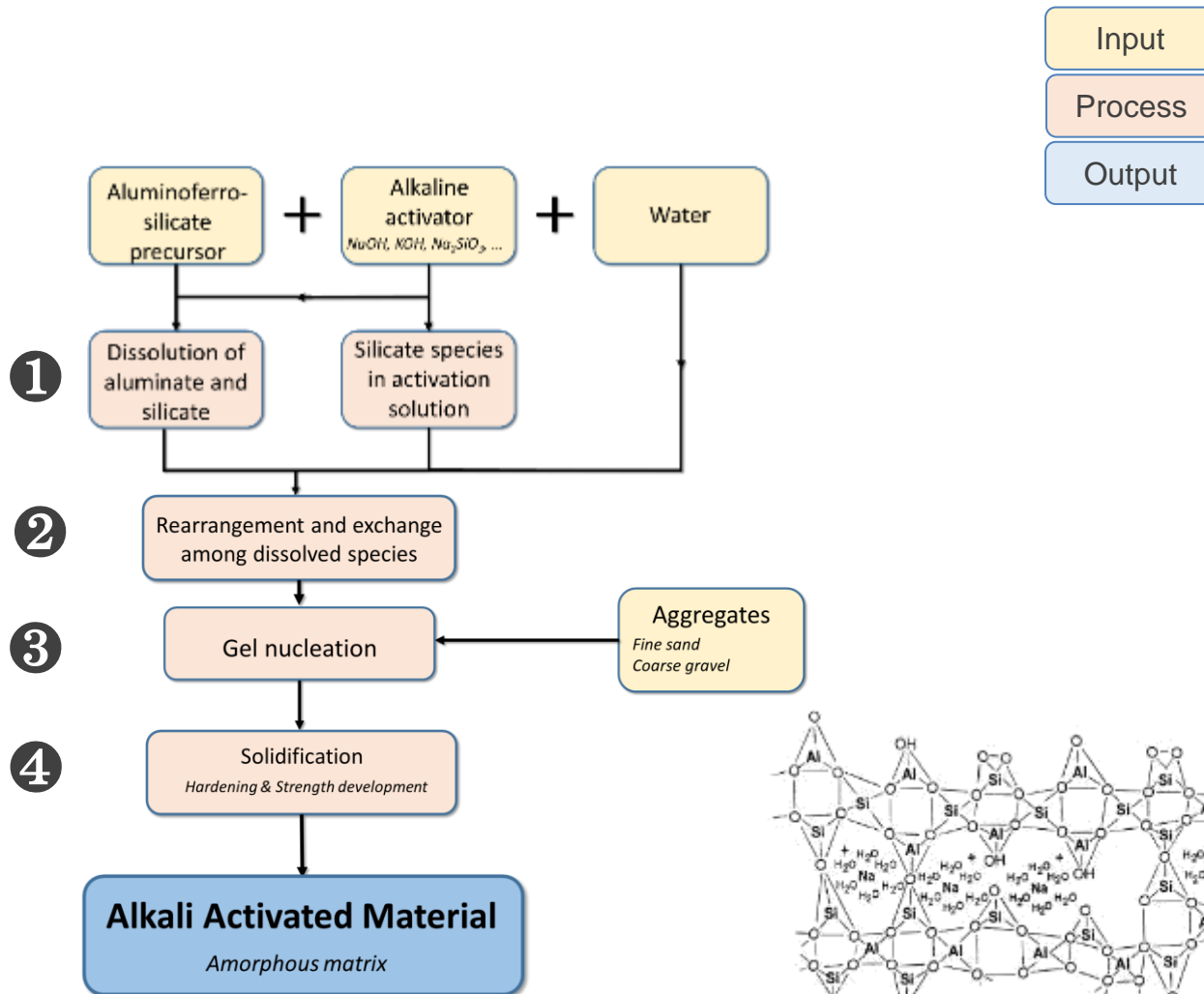
Adapted from Deventer (2007)

AAM



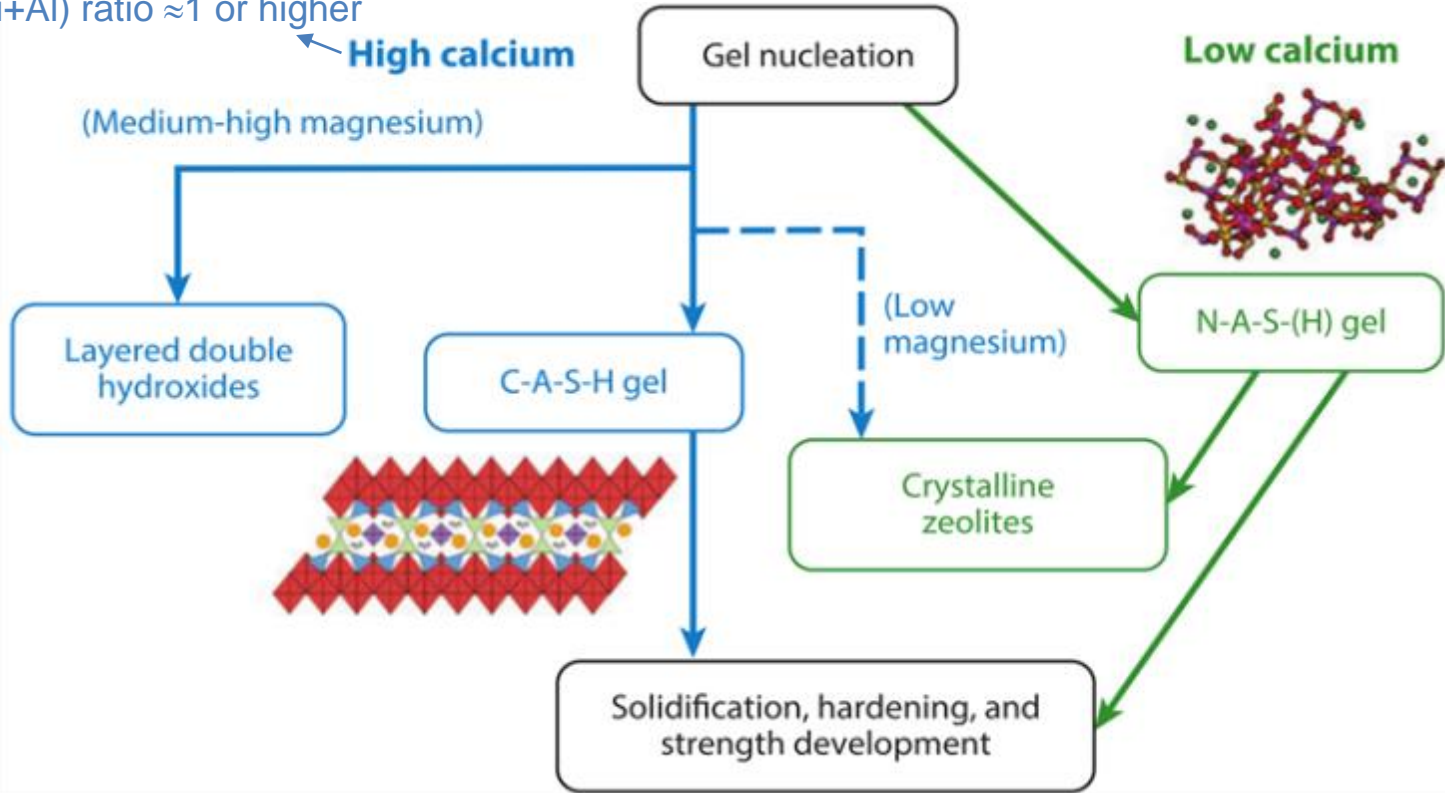
Adapted from Rowles (2008)

Alkali activated materials (AAMs) - chemistry



Alkali activated materials (AAMs) - chemistry

Ca/(Si+Al) ratio ≈ 1 or higher



Calcium (sodium) aluminosilicate hydrate (C-(N)-A-S-H)-type gel.

Low permeability

Alkali (sodium/potassium) aluminosilicate hydrate (N-A-S-H or K-A-S-H)

Excellent chemical and thermal resistance

Blended binder systems: can offer benefits of two gels

Immobilisation of Cs and Sr in alkali activated materials



ASTM C1220-98 (*Static monolithic leaching test*)

“Clean” precursor development

Start from pure Al, Si and Ca oxides

Mix and melt at high temperature (1630 °C)

Quench in water

Model AAM-composition

Different compositions

2 M NaOH

1 wt% Cs ; 0.1 wt% Sr

Leaching **28 d** at 90 °C

- [PhD Niels Vandevenne, 2019]

Immobilisation of Cs and Sr in alkali activated materials



ASTM C1220-98 (*Static monolithic leaching test*)

Cs

92 % immobilisation 28 d leaching 90 °C

Dependent on the **Si/Al** and **Ca/(Si+Al)** ratios of the precursor

Combination of initial **wash-off**, **diffusion** and **depletion** of an **easily-leachable fraction**

Sr

99 % immobilisation 28 d leaching 90 °C

Dependent on the **Ca/(Si+Al)** ratio

Limited to a **small fraction** present on or near the **surface**

Immobilisation of Cs and Sr in alkali activated materials



- **Control** of the matrix?
 - Residues \leftrightarrow model compounds
 - Variations in the (trace) element concentration?
 - Impact of the **waste-loading**?
 - Using **lower alkaline** activating solutions
 - **Durability** studies for use in long-term waste management

Effect of waste-loading on matrix properties

Cs^+ addition: Homogeneous incorporation in AAM
No significant effects on
Early reaction kinetics

Sr^{2+} addition: Sr-precipitation ($\text{Sr}(\text{OH})_2$)
Significant effects on
Early reaction kinetics

- [PhD Niels Vandevenne, 2019]

Effects of ionizing radiation on concrete



Dehydration

Radiolysis

Decomposition of water and formation of H₂ gas

Heating

Evaporation of water

Radiation-induced carbonation

Radiolysis

Formation of H₂O₂

→ CaO₂.H₂O
→ Ca(OH)
→ CaCO₃

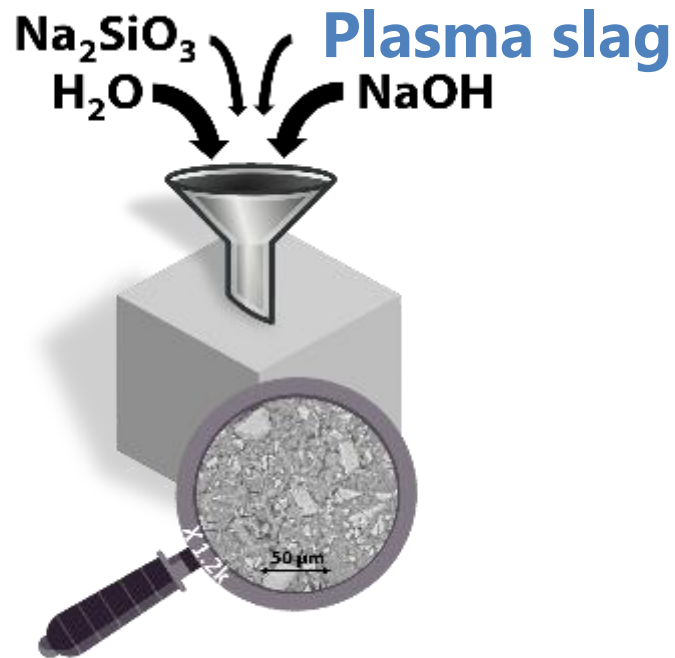
Drying & Shrinkage

Decomposition

Cracking & Loss of strength

- [PhD Bram Mast, 2020]

Radiation-induced mechanical degradation in AAMs



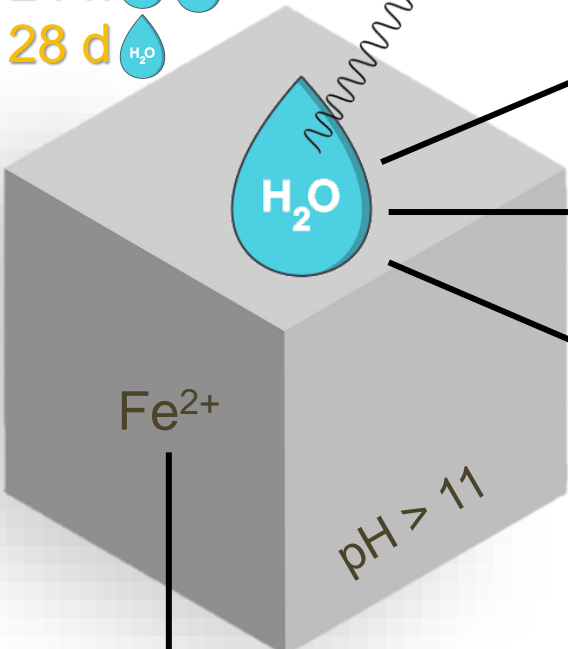
IP composite	wt.%
SiO_2	34.0
Fe_xO_y^*	24.1
CaO	21.1
Al_2O_3	11.9
Na_2O	6.3
Other	2.1

* 57% FeO and 43% Fe_2O_3

- Design4application
 - **Radioactive waste encapsulation**
 - Less $\text{Ca}(\text{OH})_2$
 - Reduced H_2O content: less damage as a result of dehydration
 - High immobilization capacity
 - **Good Shielding properties**

Effects of gamma irradiation on curing of Fe-rich AAMs

1 h
24 h
28 d



H₂

H₂O₂: strong oxidant

O₂ acts as catalyst for $\bullet e_{aq}^-$ and $\bullet OH$ removal

Fe²⁺ oxidation

Strength Increase

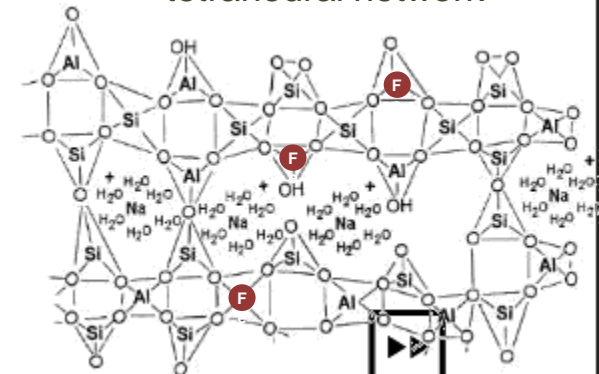
Formation of FeO(OH) nanoparticles

Nucleation sites for polymerization

Fe³⁺ incorporation in tetrahedral network

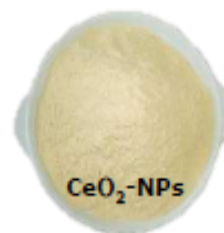
Very strong oxidants

Fenton Reactions:



Immobilisation of liquid nuclear waste

1. Adsorption on **nanoparticles**



Cerium Oxide
Nanoparticles
(CeO₂ NPs)



2. Final immobilisation with **new cementitious binders**



Collaborative Doctoral Partnership with JRC (Joint Research Centre)

- [PhD Angela Mooren]



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Lessons learned



- AAMs have the potential to outperform OPC but a lot of extra studies are required.
- **Availability** and **variations** in elemental composition of input materials
→ issues for **controlled** nuclear waste encapsulation



- **Presence of iron:** additional effects of gamma irradiation on curing of Fe-rich AAMs
- Not the best idea to use (long lived) NORM residues for immobilization of (short lived) Cat A waste